



# higher education & training

Department:  
Higher Education and Training  
**REPUBLIC OF SOUTH AFRICA**

T1390(E)(D1)T

**NATIONAL CERTIFICATE**

**POWER MACHINES N5**

(8190035)

**1 December 2017 (X-Paper)**

**09:00–12:00**

**REQUIREMENTS:** Steam Tables (BOE 173).  
Superheated Steam Tables (Appendix to BOE 173)

Candidates need drawing instruments.  
Calculators may be used.

**This question paper consists of 5 pages and 3 formula sheets.**

**DEPARTMENT OF HIGHER EDUCATION AND TRAINING**  
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NATIONAL CERTIFICATE  
POWER MACHINES N5  
TIME: 3 HOURS  
MARKS: 100

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**INSTRUCTIONS AND INFORMATION**

1. Answer ALL the questions.
  2. Read ALL the questions carefully.
  3. Number the answers according to the numbering system used in this question paper.
  4. Write neatly and legibly.
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**QUESTION 1**

- 1.1 State THREE compounds with their chemical symbols that are formed when the combustion of fuel is complete. (3 × 2) (6)
- 1.2 State TWO functions of a steam condenser. (2 × 2) (4)
- [10]**

**QUESTION 2**

Air in a cylinder has a pressure of 95 kPa and a temperature of 38 °C. The air is compressed according to the law  $PV^{1,24} = C$ , until the pressure is 650 kPa. The initial volume of air is 1,25 m<sup>3</sup>. Take the gas constant R as 0,287 kJ/kg.

Calculate:

- 2.1 The mass of the air (4)
- 2.2 The final volume of the air after compression (3)
- 2.3 The final absolute temperature of the air after compression (4)
- 2.4 The work done on the air during compression (4)
- [15]**

**QUESTION 3**

Feedwater enters an economiser at a temperature of 32,9 °C in a steam boiler plant. The boiler has a pressure of 2 500 kPa and uses 650 kg of coal per hour. The calorific value of the coal is 33 MJ/kg. Feedwater from the economiser enters the boiler at a temperature of 90 °C. The boiler produces 5 000 kg of wet steam per hour and the steam is 0,95 dry.

Calculate:

- 3.1 The quantity of heat absorbed by the feedwater in the economiser (4)
- 3.2 The thermal efficiency of the boiler (7)
- 3.3 The equivalent evaporation (EE) from and at 100 °C (4)
- [15]**

**QUESTION 4**

The barometer and manometer readings of a surface condenser are 755 mmHg and 650 mmHg respectively. The condenser receives 200 kg of steam per minute and is 10% wet. The inlet temperature of the cooling water is 15 °C and the outlet temperature is 23 °C. The condensate leaves the condenser at a temperature of 45,8 °C.

Take specific heat capacity of water as 4,187 kJ/kg.K.

Calculate:

- 4.1 The absolute pressure (3)
- 4.2 The mass of cooling water (9)
- 4.3 The logarithmic temperature difference (3)
- [15]**

**QUESTION 5**

A liquid fuel  $C_7H_{18}$  of 1 kg mass with 40% excess air is completely burned. The atomic masses of these elements are:

Hydrogen = 1  
Oxygen = 16  
Carbon = 12

Assume that air consists of 23% oxygen and 77% nitrogen by mass.

Calculate:

- 5.1 The theoretical mass of air required for complete combustion of 1 kg of the fuel (5)
- 5.2 The actual mass of air required for the combustion of 1 kg of the fuel (2)
- 5.3 The mass of each product of combustion (4)
- 5.4 The percentage of each mass of combustion (4)
- [15]**

**QUESTION 6**

1 kg mass of air enters a single cylinder, single-stage air compressor at a pressure of 103 kPa and a temperature of 22 °C. The air is then compressed according to the law  $PV^n = C$  to a pressure of 900 kPa and a volume which is  $\frac{1}{6}$  of the initial volume. Ignore clearance volume. Take R for air as 0,287 kJ/kg.K.

Calculate:

- |     |                                      |             |
|-----|--------------------------------------|-------------|
| 6.1 | The compression index (n)            | (5)         |
| 6.2 | The absolute temperature compression | (3)         |
| 6.3 | The volume after compression         | (3)         |
| 6.4 | The work done per kg of air          | (4)         |
|     |                                      | <b>[15]</b> |

**QUESTION 7**

The velocity of a gas leaving the nozzle of a single-stage impulse turbine is 900 m/s and the nozzle angle is 20°. The blade velocity is 300 m/s and the blade velocity of the coefficient is 0,7. The mass flow rate of the gas is 1 kg/s. The inlet and the outlet angles of the moving blade are equal.

- |       |  |             |
|-------|--|-------------|
| 7.1   | Construct a velocity diagram in the ANSWER BOOK (landscape) and enter ALL the values (m/s) onto the diagram. Use the scale 1 mm : 5 m/s. |             |
|       | NOTE: NO marks will be given if the values are not indicated on the diagram and if the diagram is NOT constructed to scale.              | (8)         |
| 7.2   | Determine the following from the velocity diagram:   |             |
| 7.2.1 | The inlet velocity of the moving blade   | (1)         |
| 7.2.2 | The outlet velocity of the moving blade  | (1)         |
| 7.2.3 | The axial thrust   | (2)         |
| 7.2.4 | The diagram efficiency   | (3)         |
|       |  | <b>[15]</b> |

**TOTAL: 100**

## POWER MACHINES N5

## FORMULA SHEET

1.  $Q = W + \Delta U$

2.  $\Delta U = mC_v\Delta T$

3.  $Q = mC_v\Delta T$

4.  $Q = mC_p\Delta T$

5.  $Q = P_1V_1 \ln \frac{V_2}{V_1}$

6.  $\Delta S = m \left( C_v \ln \frac{T_2}{T_1} + R \ln \frac{V_2}{V_1} \right)$

7.  $W = P_1\Delta V$

8.  $W = P_1V_1 \ln \frac{V_2}{V_1}$

9.  $W = \frac{P_1V_1 - P_2V_2}{n-1}$

10.  $W = \frac{P_1V_1 - P_2V_2}{\gamma-1}$

11.  $R = C_p - C_v$

12.  $\gamma = \frac{C_p}{C_v}$

13.  $\frac{P_1V_1}{T_1} = \frac{P_2V_2}{T_2}$

14.  $PV = mRT$

15.  $P_1V_1 = P_2V_2$

16.  $P_1V_1^n = P_2V_2^n$

17.  $\frac{T_2}{T_1} = \left( \frac{P_2}{P_1} \right)^{\frac{n-1}{n}} = \left( \frac{V_1}{V_2} \right)^{n-1}$

18.  $\frac{T_2}{T_1} = \left( \frac{P_2}{P_1} \right)^{\frac{\gamma-1}{\gamma}} = \left( \frac{V_1}{V_2} \right)^{\gamma-1}$

19.  $h = h_f + \chi h_{fg}$

20.  $h = h_g + C_p\Delta T$

21.  $h = h_f + h_{fg} = h_g$

22.  $V_{\text{sup}} = \frac{n-1}{n} \left( \frac{h_{\text{sup}} - 1941}{P} \right)$

23.  $\chi = \frac{V_m}{V_g}$

24.  $\chi = \frac{M}{M + m}$

25.  $U = H - PV$

26.  $gZ_1 + U_1 + P_1V_1 + \frac{1}{2}C_1^2 + Q =$

$gZ_2 + U_2 + P_2V_2 + \frac{1}{2}C_2^2 + W$

27.  $\eta = \frac{\dot{m}_s(h_2 - h_1)}{\dot{m}_f CV}$

28.  $EE = \frac{\dot{m}_s(h_2 - h_1)}{\dot{m}_f 2257}$

29.  $p = (B_m \pm M_m) \frac{101,325}{760}$

$$30. \quad m = \frac{100}{23} \left[ C \frac{8}{3} + 8H_2 + S - O_2 \right]$$

$$31. \quad C_x H_y + \left( x + \frac{y}{4} \right) O_2 = xCO_2 + \frac{y}{2} H_2O$$

$$32. \quad H.C.V. = (CV_C \cdot C) + CV_{H_2} \left( H_2 - \frac{O_2}{8} \right) + (CV_S \cdot S)$$

$$33. \quad L.C.V. = H.C.V. - h_{fg} (9H_2)$$

$$34. \quad H.C.V. = \frac{(m_w + m_e) C_p \Delta T}{m_f}$$

$$35. \quad W = P_1 V_e \left( \frac{n}{n-1} \right) \left[ \left( \frac{P_2}{P_1} \right)^{\frac{n-1}{n}} - 1 \right] = mRT_1 \left( \frac{n}{n-1} \right) \left[ \left( \frac{P_2}{P_1} \right)^{\frac{n-1}{n}} - 1 \right]$$

$$36. \quad \eta_c = \frac{V_e}{V_s} \cdot 100 = 1 - \frac{V_c}{V_s} \left[ \left( \frac{P_2}{P_1} \right)^{\frac{1}{n}} - 1 \right] = 1 + \alpha - \alpha (r_p)^{\frac{1}{n}}$$

$$37. \quad \eta_a = \frac{V_a}{V_s} \cdot 100$$

$$38. \quad \dot{F}_c = \dot{m}(C_{fe} - C_{fi})$$

$$39. \quad \dot{P} = \dot{m}U[C_{wi} - (-C_{we})]$$

$$40. \quad \eta = \frac{2U[C_{wi} - (-C_{we})]}{C_{ai}} \cdot 100$$

$$41. \quad U = \pi DN$$

$$42. \quad \dot{m}V = AC$$

$$43. \quad (m + M)g = m\omega^2 h$$

$$44. \quad V_s = \frac{\pi}{4} D^2 L$$

$$45. \quad \theta_1 = t_c - twi$$

$$46. \quad \theta_2 = t_c - two$$

$$47. \quad \text{Log. temp. diff.} = \frac{\theta_1 - \theta_2}{\ln \frac{\theta_1}{\theta_2}}$$

$$48. \quad P_{iso} = P_1 V_1 \ln \left( \frac{P_2}{P_1} \right)$$

$$49. \quad P_{act} = \frac{n}{n-1} P_1 V_1 \left[ \left( \frac{P_2}{P_1} \right)^{\frac{n-1}{n}} - 1 \right]$$

$$50. \quad N_{iso} = \frac{P_{iso}}{P_{act}} \cdot 100$$